

FORM PTO-1390
(REV. 5-93)

U.S. DEPARTMENT OF COMMERCE
PATENT AND TRADEMARK OFFICE

JCO7 Rec'd PCT/PTO 15 FEB 2002

ATTORNEY'S DOCKET NUMBER

6670/OK215

**TRANSMITTAL LETTER TO THE UNITED STATES
DESIGNATED/ELECTED OFFICE (DO/EO/US)**

10/069035

INTERNATIONAL APPLICATION NO.
PCT/CA00/00865

INTERNATIONAL FILING DATE
July 21, 2000

PRIORITY DATE CLAIMED
August 18, 1999

TITLE OF INVENTION

METAL HYDRIDE BANDS AND USE THEREOF FOR STORING HYDROGEN

APPLICANT(S) FOR DO/EO/US

Robert SCHULZ; Sabin BOILY; René DUBUC; Marco BLOUIN; Guy LALANDE

Applicant herewith submits to the United States Designated/Elected office (DO/EO/US) the following items and other information:

1. ☒ This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
2. ☐ This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
3. ☒ This is an express request to begin national examination procedures (35 U.S.C. 371 (f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S. C. 371 (b) and PCT Articles 22 and 39 (1).
4. ☒ A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date.
5. ☒ A copy of the International Application as filed (35 U.S. C. 371 (c) (2))
 - a. ☐ is transmitted herewith (required only if not transmitted by the International Bureau).
 - b. ☒ has been transmitted by the International Bureau
 - c. ☐ is not required, as the application was filed in the United States Receiving Office (RO/US)
6. ☒ A translation of the International Application into English (35 U.S. C. 371 (c)(2)).
7. ☐ Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371 (c)(3))
 - a. ☐ are transmitted herewith (required only if not transmitted by the International Bureau).
 - b. ☒ have been transmitted by the International Bureau.
 - c. ☐ have not been made; however, the time limit for making such amendments has NOT expired.
 - d. ☐ have not been made and will not be made.
8. ☐ A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371 (c) (3)).
9. ☒ An oath or declaration of the inventor(s) (35 U.S.C. 371(c)(4)). (Unexecuted)
10. ☐ A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371 (c)(5)).

Items 11. to 16. below concern other document(s) or information included:

11. ☒ An Information Disclosure Statement under 37 CFR 1.97 and 1.98 (with 4 references).
12. ☐ An assignment document for recording. A **separate** cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
13. ☒ A **FIRST** preliminary amendment.
☐ A **SECOND** or **SUBSEQUENT** preliminary amendment.
14. ☒ A substitute specification.
15. ☐ A change of power of attorney an/or address letter.
16. ☒ Other items or information:
Statement that Substitute Specification Contains No New Matter
Transmittal of Substitute Specification

EXPRESS MAIL CERTIFICATE

Date 2/15/02 Label No. 29137720US

I hereby certify that, on the date indicated above, this paper or fee was deposited with the U.S. Postal Service & that it was addressed for delivery to the Assistant Commissioner for Patents, Washington, DC 20231 by "Express Mail Post Office to Addressee" service.

Name (Print)

Signature

U.S. APPLICATION NO. (if known sec 37 CFR 1.50)

INTERNATIONAL APPLICATION NO. PCT/CA90/00865

Attorney's Docket Number
6670/OK215

17. [x] The following fees are submitted:

Basic National Fee (37 CFR 1.492 (a)(1)-(5)):

Search Report has been prepared by the EPO [X] or JPO []

\$890.00

International preliminary examination fee paid to USPTO (37 CFR 1.482)

\$710.00

No international preliminary examination fee paid to USPTO (37 CFR 1.482)
but international search fee paid to USPTO (37 CFR 1.445 (a) (2))...

\$740.00

Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO.....

\$1,040.00

International preliminary examination fee paid to USPTO (37 CFR 1.482)
and all claims satisfied provisions of PCT Article 33(2)-(4)....

\$100.00

ENTER APPROPRIATE BASIC FEE AMOUNT =

\$890.00

Surcharge of \$130.00 for furnishing the oath or declaration later than [] 20 [] 30
months from the earliest claimed priority date (37 CFR 1.492(e)).

\$

Claims	Number Filed	Number Extra	Rate		
Total Claims	20-20		X \$18.00	\$00	
Independent Claims	1-3		X \$80.00	\$00	

Multiple dependent claims(s) (if applicable)

+ 280

\$280.00

TOTAL OF ABOVE CALCULATIONS =

\$890.00

Reduction by 1/2 for filing by small entity, if applicable.

\$890.00

SUBTOTAL =

\$890.00

Processing fee of \$130.00 for furnishing the English translation later the [] 20 [] 39
months from the earliest claimed priority date (37 CFR 1.492(f)).

+

\$

TOTAL NATIONAL FEE =

\$890.00

Fee for recording the enclosed assignment (37 CFR 1.21(h)). the assignment must be accompanied by an
appropriate cover sheet (37 CFR 3.28, 3.31). \$40.00 per property

+

\$0.00

TOTAL FEES ENCLOSED =

\$890.00

Amount to be:
refunded

\$

charged

\$

a. [X] A check in the amount of \$890.00 to cover the above fees is enclosed.

b. [] Please charge my Deposit Account No.04-0100 in the amount of \$ to cover the above fees.

c. [X] The Commissioner is hereby authorized to charge any additional fees which may be required, or credit any overpayment to Deposit
Account No. 04-0100. A duplicate copy of this sheet is enclosed.

NOTE: Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed
and granted to restore the application to pending status.

SEND ALL CORRESPONDENCE TO

John C. Todaro
Darby & Darby P.C.
805 Third Avenue
New York, New York 10022-7513

SIGNATURE

John C. Todaro

NAME John C. Todaro

REGISTRATION NO. 36,036

EXPRESS MAIL CERTIFICATE

Date 2/15/02 EV 039137720US
Label No.

I hereby certify that, on the date indicated above, this paper or fee was deposited with the U.S. Postal Service & that it was addressed for delivery to the Assistant Commissioner for Patents, Washington, DC 20231 by "Express Mail Post Office to Addressee" service.

PLEASE CHARGE ANY DEFICIENCY UP TO \$300.00 OR CREDIT ANY EXCESS IN THE FEES DUE WITH THIS DOCUMENT TO OUR DEPOSIT ACCOUNT NO. 04-0100

Name (Print)

Signature

Customer No.:



07278

PATENT TRADEMARK OFFICE

Docket No: 6670/OK215

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Application of: Robert SCHULZ et al.

Serial No: T/B/A (National Phase of PCT/CA00/00865)

Filed: Concurrently Herewith

For: METAL HYDRIDE BANDS AND USE THEREOF FOR STORING HYDROGEN

Hon. Commissioner for Patents
Washington, DC 20231

PRELIMINARY AMENDMENT

Sir:

Prior to examination, please amend the above-identified application as follows:

In the Claims:

Cancel claims 1-20 of the substitute specification without prejudice and substitute therefor:

21. A piece comprising at least one metal hydride capable of absorbing hydrogen in a reversible manner, said piece being in the form of a thin and dense strip obtained by rolling a powder of said at least one metal hydride, said strip being obtained at a temperature lower than 400°C.

22. The piece according to claim 21 further comprising a first additional component for heat-related actions selected from the group consisting of supplying heat or evacuating heat.

23. The piece according to claim 22, wherein said first additional component is a binder for the powder of said at least one metal hydride.

24. The piece according to claim 21 further comprising a second additional component which is a binder for the powder of said at least one metal hydride.

25. The piece according to claim 24, wherein at least one of said first and said second additional components is in the form of a powder additive.

26. The piece according to claim 25, wherein said powder additive comprises Mg.

27. The piece according to claim 24, wherein at least one of said first and said second additional components is in the form of a tridimensional matrix that is rolled together with the powder of said at least one metal hydride.

28. The piece according to claim 27, said matrix comprising metal and having a porous structure.

29. The piece according to claim 24, wherein at least one of said first and said second additional components is in the form of a plate in direct contact with the powder of said at least one metal hydride.

30. The piece according to claim 24, wherein at least one of said first and said second additional components is in the form of a plate in direct contact with a tube selected from the group consisting of a tube containing said at least one metal hydride and a rolled tube containing said powder of said at least one metal hydride.

31. The piece according to claim 24, wherein at least one of said first and said second additional components are up to 50% of the weight of the whole piece.

32. The piece according to claim 31, wherein at least one of said first and said second additional components are up to 30% of the weight of the whole piece.

33. The piece according to claim 21, said strip having a thickness equal to or less than 1 mm.

34. The piece according to claim 21 wherein said powder of said at least one metal hydride is nanocrystalline.

35. The piece according to claim 34 wherein said at least one nanocrystalline metal hydride comprises MgH_2 -5at. %.

36. The piece according to claim 21 wherein said strip is of a shape selected from the group consisting of straight, stacked, folded, spiral, curved, twisted and cut shapes.

37. The piece according to claim 21, said piece being formed to have intrinsic electric characteristics allowing measurement of its hydrogen content.

38. The piece according to claim 21, said piece being formed to have intrinsic electric characteristics allowing desorption of hydrogen by circulation of an electrical current.

39. A method for storing and transporting hydrogen in a tank, comprising use of a piece according to claim 21.

40. A method for stocking and transporting energy in a battery of the Ni-MH type, comprising use of a piece according to claim 21.

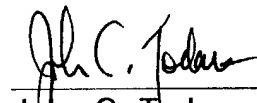
REMARKS

Claims 1-20 of the PCT application (as presented in the substitute specification filed herewith) have been cancelled and rewritten to be placed in better form for U.S. practice, including the removal of the multiple dependent claims. No new matter is introduced by the amended claims.

An early and favorable examination on the merits of this application is earnestly solicited.

Respectfully submitted,

Dated: February 15, 2002



John C. Todaro
Reg. No. 36,036
Attorney for Applicant(s)

DARBY & DARBY P.C.
805 Third Avenue
New York, NY 10022
212-527-7700

Serial No. T/B/A
(National Phase of PCT/CA00/00865)

Docket No. 6670/OK215

EXPRESS MAIL CERTIFICATE

Date 2/15/02 Label No. EV039137720Us

I hereby certify that, on the date indicated above, this paper or fee was deposited with the U.S. Postal Service & that it was addressed for delivery to the Assistant Commissioner for Patents, Washington, DC 20231 by "Express Mail Post Office to Addressee" service.

PLEASE CHARGE ANY DEFICIENCY UP TO \$300.00 OR CREDIT ANY EXCESS IN THE FEES DUE WITH THIS DOCUMENT TO OUR DEPOSIT ACCOUNT NO. 04-0100

DBPeeK [Signature]
Name (Print) Signature

Customer No.:



07278

PATENT TRADEMARK OFFICE

Docket No: 6670/OK215

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Application of: Robert SCHULZ et al.

Serial No: T/B/A (National Phase of PCT/CA00/00865)

Filed: Concurrently Herewith

For: METAL HYDRIDE BANDS AND USE THEREOF FOR STORING HYDROGEN

Hon. Commissioner for Patents
Washington, DC 20231


TRANSMITTAL OF SUBSTITUTE SPECIFICATION (37 C.F.R. § 1.125)

Sir:

1. Enclosed at Tab A is a substitute specification for the originally filed specification in International Application No. PCT/CA00/00865, of which this application is a national phase entry under 35 U.S.C. § 371.
2. This substitute specification is being voluntarily submitted, in order to facilitate the processing of the application.

3. Also enclosed at Tab B is a marked-up copy of the substitute specification showing the matter being added to (underlined) and the matter being deleted from (in brackets) the specification.
4. Accompanying this transmittal is a statement, as required by 37 C.F.R. § 1.125, that the substitute specification transmitted herewith contains no new matter.

Dated: February 15, 2002


John C. Todaro
Reg. No. 36,036
Attorney for Applicant(s)

DARBY & DARBY P.C.
805 Third Avenue
New York, NY 10022
212-527-7700

**METAL HYDRIDE BANDS AND USE THEREOF
 FOR STORING HYDROGEN**

5

BACKGROUND OF THE INVENTION

The present invention relates to pieces made of at least one metal hydride, which are in the form of strips obtained by rolling.

10 The invention also relates to the use of these pieces for the storage and transportation of hydrogen in hydrogen storage tanks.

The invention further relates to the use of the same pieces in batteries of the Ni-MH type and in equipment used for purification, separation, compression, detection, cooling, heating, storing and generation of energy.

15 In the present description and the appended claims, the expression "metal hydrides" is used to designate all metals, alloys, composites and other materials known to absorb hydrogen in a reversible manner, whether these products are in a non-hydrogenated form (that is before absorption or after desorption of hydrogen) or in a hydrogenated form (that is after absorption and
 20 before desorption of hydrogen). These hydrides can be of the type known to be "high temperature". By way of examples of such hydrides, mention can be made to Na, Mg, Mg₂Ni, Li, Ti, Zr and Ca. These hydrides may also be of the type "low or relatively low temperature". By way of examples of such hydrides, mention can be made to MmNi₅, LaNi₅, CaNi₅, FeTi, Ti_{0.98} Zr_{0.02} V_{0.43} Fe_{0.09} Cr_{0.051} Mn_{1.5} and
 25 the Bogdanovic alloys.

BRIEF DESCRIPTION OF THE PRIOR ART

The use of metal hydrides for storing and transporting hydrogen is well known *per se*. The hydrogen which is so stored is usually used as a source of
 30 energy for hydrogen powered vehicles, batteries or machines or other kind of

equipments used for purification, separation, compression, detection, cooling and heating.

If all these potential applications are extremely interesting, there is still a practical problem of heat and mass (hydrogen), transfer which is associated to the use of metal hydrides having high hydrogen absorption and desorption kinetics. Indeed, absorption of hydrogen by a metal hydride is an exothermic phenomenon. Therefore, it is necessary to quickly evacuate the heat during the absorption, since otherwise the process will be slowed down. On the other hand, desorption is an endothermal phenomenon. It is therefore necessary to provide heat in a quick manner in order to extract hydrogen at a high rate from a metal hydride. When one wants to hydrogenate or dehydrogenate very quickly a material and thus, at the same time, to extract or supply heat, the thermal and diffusion paths must be reduced as much as possible, thereby making it necessary for the material to have at least one dimension which is small. Moreover, in order to obtain a high storage capacity per volume unit, it is necessary to consolidate and densify the hydride when the latter is in the form of a powder.

In order to tentatively overcome this problem, Japanese patent application No. JP-A-60/262.830 published on December 26, 1985 in the name of TOYOTA CENTER OF RESEARCH & DEVELOPMENT LAB suggests to give the shape of a thin sheet to the hydride. This sheet is obtained by "molding" the hydride around a metal mesh by means of a synthetic resin, such as a silicon resin. Such a "molding" process is quite difficult to carry out from an industrial standpoint. Moreover, as everybody knows, synthetic resins (polymers) cannot be exposed to high temperatures. Accordingly, the hydrides that can be used can only be low temperature metal hydrides (example 1 of this laid-open application refers to LaNi_5). It is impossible to use this technology with high temperature metal hydrides. Moreover, since synthetic resins do not conduct electricity and heat, it is necessary to incorporate heat pipes into the molded sheet in order to circulate a heating/cooling fluid for heat exchange.

OBJECTS AND SUMMARY OF THE INVENTION

A first object of the present invention is to give to a metal hydride a form suitable to efficiently solve the practical problem of heat and mass transfers
5 mentioned hereinabove.

A second object of the invention is to achieve the requested forming of a metal hydride by using a technology that can easily be carried out in mass production and at low cost.

A third object of the invention is to generalize the use of metal
10 hydrides that are so formed to all possible industrial applications for this type of product known to be capable of absorbing hydrogen in reversible manner.

In accordance with the invention, these various objects are achieved with at least one piece made of at least one metal hydride capable of absorbing hydrogen in a reversible manner, said piece being in the form of a thin and dense
15 strip obtained by rolling of a powder of said at least one metal hydride and being characterized in that said strip is obtained at a temperature lower than 400°C.

This piece preferably has a thickness equal to or lower than 1 mm and made from a powder metal hydride(s) of nanocrystalline structure (crystallites of a size lower than 100 nm). This piece can have any shape: straight, stacked,
20 folded, spiral, curved, twisted or cut.

In accordance with a particularly preferred embodiment of the invention, this piece may contain a first additional component for supplying and/or evacuating heat. This first additional component may also act as a binder for the powder of metal hydride(s).

25 The piece may also contain a second additional component acting as a binder for the powder of metal hydride(s).

The first and/or second additional components are preferably in the form of a powder additive. This additive may be a powder of Mg, Cu or Al.

Alternatively, the first and/or second additional components may be in the form of a tridimensional matrix preferably made of metal and having porous structure, that is rolled together with the powder of metal hydride(s).

The first and/or second additional components may also be in the form of a plate or tube in direct contact with the powder of metal hydride(s).

Preferably, the weight of the first and/or second additional components corresponds to a maximum of up to 50% of the weight of the whole piece. More preferably, this weight does not exceed 30% only of the weight of the whole piece.

Intrinsically, the piece according to the invention is capable of absorbing hydrogen in a reversible manner. It can be formed to have intrinsic electric characteristics allowing measurement of its hydrogen content. It can also be formed to have intrinsic electric characteristics allowing desorption of hydrogen by circulation of an electric current.

As indicated hereinabove, the piece made of metal hydride(s) according to the invention is manufactured by rolling. This process is particularly advantageous in that it allows the manufacture of very dense strips of a very small thickness which as a result, have the above-mentioned characteristics. This process also has the advantage of being perfectly adapted for industrial production on a large scale and at low cost.

Because of its property, the piece according to the invention is particularly well adapted for use as basic components in a tank for the storage and transportation of hydrogen. It can also be used in a battery of the type Ni-MH for the storage and transportation of energy. It can further be used in other equipment selected amongst the purification, separation, compression, detection, cooling, heating, storing and generation of energy.

The invention and its advantages will be better understood upon reading the following more detailed but non-restrictive description made in reference to the accompanying drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a schematic representation of the manufacture of a piece of metal hydride according to the invention, incorporating a tridimensional porous metal structure;

5 Figure 2 is a representation of a piece of metal hydride according to the invention, associated with, on the one hand, a metal plate used as a heat carrier (and possibly as a contact electrode) and, on the other hand, mattresses of porous fibers for the supply and removal of hydrogen and for volume compensation;

10 Figure 3 is a cross-sectional view of an ensemble consisting of a piece according to the invention associated with, on the one hand, a metal plate acting as a heat carrier and, on the other hand, an insulating mattress, said figure giving a numerical simulation of the heat transfer in this ensemble during hydrogen absorption;

15 Figure 4 is a curve illustrating the variation of resistivity of a consolidated piece of a metal hydride of the formula $MgH_2-5at.\%V$ during hydrogen desorption and the variation of differential pressure of the hydrogen titration apparatus, as a function of the temperature;

20 Figure 5 is a schematic cross-sectional view of a tank for storing hydrogen, comprising a stack of pieces according to the invention;

 Figure 6 is an open perspective view of the tank shown in cross-section in Figure 5; and

 Figures 7a to 7b are perspective views showing in a schematic manner, different shapes that may have the pieces according to the invention.

25

DETAILED DESCRIPTION OF THE INVENTION

As previously indicated, the invention essentially relates to a piece made of one or more metal hydrides that are capable of absorbing hydrogen in a reversible manner. The piece is in the form of a thin and dense strip having a
30 thickness preferably equal to or lower than 1 mm and a porosity preferably lower

than 20% which is obtained by rolling a powder of the selected hydride(s) with or without additional component(s). This strip has a typical width which is preferably lower than 15 cm and a length that can be of several meters.

By way of non-restrictive examples of hydrides that can be used for the manufacture of the pieces, mention can be made to those of nanocrystalline structures disclosed in U.S. patent Nos. 5,763,363; 5,837,030; 5,872,074 and 5,906,792 either in the name of the Applicant alone or in the name of the Applicant and the MCGILL UNIVERSITY. Even more particularly mention can be made to Mg.-based hydrides.

The key feature of the invention lies in the use of the well known method of "rolling" for giving to the powder of hydrides the form of thin and dense strips and thus for allowing optimization of the transfers of mass (hydrogen) and heat involved during the process of absorption/desorption of hydrogen.

In order to obtain by rolling strips having a high structural integrity, additional components such as binders can be added as additives to the hydrides. By way of examples of such additives, reference can be made to Mg, Al, V, Cu, Li, Fe, Nb, C, graphite, Co, Ni, Mn, Cr, Ti, Zr and FeTi. These additives are preferably in the form of a powder that can be mixed with the powder of hydride(s) and rolled together with it.

As an additional component, it may also be suggested to use a tridimensional structure preferably porous and made of metal, such as a metal foam, to insert the metal hydride powder into the pores of the structure and to roll the whole to obtain a thin strip. The metal structure that is so incorporated to the piece then permits to more quickly remove heat from the strip during hydrogen absorption.

Figure 1 is illustrative of this process of manufacture. As is shown, the tridimensional metal structure 3 is introduced with a carrier plate 5 preferably made of a heat conductive material, between the rollers 7 of a rolling machine. Just before such an introduction, the powder of metal hydride 9 stored into a

tank 11 is poured into the structure 3 to fill up its pores. After rolling, the so obtained piece in the form of a thin and dense strip 1 can then be combined with other elements such as an electrical contact grid 13 and mattresses 15 of micro glass-fibers in order to obtain an operating unit ready to be incorporated into a tank for the storage of hydrogen.

In use, the rolling step can be carried out at ambient temperature or at elevated temperature (lower than 400°C). The rolls 7 can be heated and their relative position can be adjusted to obtain a requested thickness and density.

As can be understood, the above-mentioned additives are used and rolled with the powder of metal hydride essentially in order to give structural integrity to the strip and to act as a heat conductor. Accordingly, it is important to preferably use additives in the form of metal powders or metal fibers, which are capable of acting not only as binders but also as heat conductors.

Together with these additives or as substituents for them, use can also be made of the tridimensional metal structure, or of the carrier plate or grid as already referred to hereinabove, or also of a metal tube into which the hydride powder is inserted before the rolling step. These components should also be made of a heat conductive material and be preferably provided with holes to let flow the hydrogen to be absorbed or desorbed.

In order to promote the desorption of hydrogen, use can be made of the intrinsic electric characteristics of the pieces in the form of strips obtained by rolling. Thus, for example, it is known that magnesium hydride MgH_2 is an insulating material. When the content of metal additives in a form of a powder or in a form of porous structure exceeds the percolation point, a strip made from this hydride becomes conductive. Moreover, during the desorption, MgH_2 switches from an insulating state to a conductive state. Thus, use can be made of these characteristics to induce desorption of hydrogen from the strip by passing an electric current therein. The change of resistivity of the pieces as a function of the content of hydrogen contained in them may also be used as an indicator of the amount of hydrogen within the tank.

Figure 4 illustrates as an example, the change of resistivity of a consolidated piece made of MgH_2 and the change of differential pressure of the hydrogen titration apparatus used for measuring the amount of hydrogen within this piece as a function of the temperature during a desorption step. This figure shows that, during desorption, magnesium hydride (MgH_2) which is an insulating material, becomes conductive (Mg). This figure can be associated to the table given below which indicates the value of the electric resistance of 1 mm thick rolled pieces made from the same nanocrystalline metal hydride of formula MgH_2 -5at%V, to which were added variable amounts of a binder consisting of Mg of granulometry ranging from $105\mu\text{m}$ to $850\mu\text{m}$ (the internal resistance of the measurement apparatus was 2000Ω).

Amount of binder (Mg) added (expressed in weight)	0	10%	15%	20%	30%
Measured resistance (Ω)	4×10^{10}	1×10^{10}	3000	< 2000	< 2000

It will be noticed that when the additional component acting as a binder is in the form of a powder that is mixed with and integrated to the hydride powder, it is of the highest importance that this component be well distributed and occupy all the volume of the piece, as is the case of a tri-dimensional metal structure when it is inserted into the hydride powder during rolling (see Figure 1). As a matter of fact, one and/or the other of these two solutions are very preferable to a mere pressing of the hydride powder onto a substrate without any additives.

The concept of using pieces in the form of rolled strips of metal hydrides prepared as disclosed hereinabove, is sufficiently flexible to be adapted to a multitude of applications. These strips may be of any shape. They may be flat and rectilinear strips, cut into straight sections (see figures 5 and 6). Alternatively, the strips may be folded and stacked upon themselves (figure 7a),

rolled in a spiral (figure 7b), curved (figure 7C) or cut, sectioned and reassembled at an angle (figure 7d).

Figures 5 and 6 illustrate a possible use of several pieces 1 in the form of straight strips into a tank 21 for the storage and transportation of hydrogen. The tank 21 comprises an external wall 23, an internal wall 25 fixed to the external wall via a corrugated sleeve 27 acting as a mechanical reinforcement and a thermal breaking. Thanks to its structure, the sleeve also defines a plurality of conducts 29 for a heat carrying fluid (water, oil or any other liquid) which is brought at 31 and used for cooling during the absorption or for heating during the desorption of hydrogen. The hydrogen circulates via a pipe 33 opening into the container which is defined by the internal wall 25 and in which the pieces 1 are stacked. Contact plates 5 perforated with holes, act as heat carriers and are connected to the internal wall to act together with it as a heat exchanger. Mattresses of glass microfibers having a porosity of, for example, 90% with pores of, for example, 0,3 microns, are positioned between the pieces as shown in figure 2, to ensure the supply and removal of hydrogen and to absorb volume dilatation or contraction.

The tank 21 disclosed hereinabove, has numerous advantages. It is safe and of easy operation. It is easy to fill and of a great capacity. It allows liberation of hydrogen upon request, and such in a very efficient and fast manner thanks to the structure of the pieces 1 and their spacing. The absorption/desorption kinetics are very fast and the storage capacity is very high. The number of cycles of absorption/desorption is also high.

Such makes the tank 21 useful in the field of transportation (cars, buses, tractors, trucks, recreation vehicles, planes, boats, trains, military transports, etc...) or for the manufacturers/users of fuel cells, the producers of decentralized energy, the users of hydrogen of very high purity and the users of filters for hydrogen.

So far, in all the applications involving hydrogen especially as a source of energy for fuel cells or internal combustion engines or as sources for the

production of hydrogen in reformers or electrocatalic machines, the storage of hydrogen was done either under very high pressure with the safety problem that such generates (pressure $>> 300$ psi), or in liquid form, with the problems of the storage at very low temperatures ($< -253^{\circ}\text{C}$). Another solution is the use of
5 conventional hydrides in the form of a non compacted powder or of massive pieces with the problems that such generate, in particular, a very low absorption/desorption kinetics and a low storage capacity.

The use of rolling as proposed according to the invention for the manufacture of thin and dense strips of hydrides, permits to optimize the
10 absorption/desorption kinetics in storage tanks and to maximize the heat transfer. The use of the intrinsic electric properties of the consolidated strips close to the percolation point, permits to measure the content of hydrogen and promote the desorption of hydrogen. Because of their structure, the so-obtained pieces permit to optimize mass and heat transfer within the tanks of metal hydrides. It provides
15 large hydrogen absorption/desorption kinetics, fast mass and heat transfer and high number of cycles of absorption/desorption. Also because of their structure, the consolidated pieces are safe as compared to a powder of non-binded hydride which presents a certain level of pyrophoricity.

The technology disclosed hereinabove and the pieces of hydride(s) in the
20 form of strips according to the invention may also be used efficiently for applications in batteries of the Ni-MH type (nickel-metal hydride).

The field of application of the hydride strips and the hydrogen storage tanks using these pieces in the form of rolled hydride strips according to the invention is therefore very wide. This technology is particularly well adapted for
25 use with nanocrystalline hydrides since these materials have very high hydrogen absorption and desorption kinetics. It is particularly well adapted for the Mg-containing hydrides (Mg, Mg_2Ni and the related materials), Li, Na, Ti, Zr and Ca, which are known to be "stable" hydrides operating at a high temperature and for which the problem of heat exchange is important. This technology may also be
30 used with low temperature hydrides such as MmNi_5 , LaNi_5 , CaNi_5 , FeTi , $\text{Ti}_{0.98}$,

Zr_{0.02}, V_{0.43}, Fe_{0.09}, Cr_{0.05}, Mn_{1.5}, Bogdanovic alloys, etc., that is, in summary, with all hydrides of the AB₅, AB₂, AB, A₂B, types, solid solutions, nanocrystalline and amorphous alloys, complex hydrides and even carbon, nanotubes, etc...

CLAIMS

1. A piece made of at least one metal hydride capable of absorbing
5 hydrogen in a reversible manner, said piece being in the form of a thin and dense
strip obtained by rolling a powder of said at least one metal hydride, characterized
in that said strip is obtained at a temperature lower than 400°C.

2. The piece according to claim 1, characterized in that it comprises a
10 first additional component for supplying and/or evacuating heat.

3. The piece according to claim 2, characterized in that the first
additional component is also acting as a binder for the powder of said at least one
metal hydride.

4. The piece according to claim 1 or 2, characterized in that it
15 comprises a second additional component acting as a binder for the powder of
said at least one metal hydride.

5. The piece according to claim 2, 3, or 4, characterized in that the
20 first and/or second additional component is in the form of a powder additive.

6. The piece according to claim 5, characterized in that the powder
additive consists of Mg.

7. The piece according to claim 2, 3 or 4, characterized in that the first
25 and/or second additional component is in the form of a tridimensional matrix that
is rolled together with the powder of said at least one metal hydride.

30

8. The piece according to claim 7, characterized in that the matrix is made of metal and has a porous structure.

5 9. The piece according to claim 2, 3 or 4, characterized in that the first and/or second additional component is in the form of a plate in direct contact with the powder of said at least one metal hydride or of a tube containing said at least one metal hydride or of a rolled tube containing said powder of said at least one metal hydride.

10 10. The piece according to any one of claims 2 to 9, wherein the first and/or second additional components have a weight amounting to up to 50% of the weight of the whole piece.

15 11. The piece according to claim 10, characterized in that the first and/or second additional components have a weight amounting to up to 30% of the weight of the whole piece.

20 12. The piece according to any one of claims 1 to 11, characterized in that the strip has a thickness equal to or lower than 1 mm.

13. The piece according to any one of claims 1 to 12, characterized in that the powder of said at least one metal hydride is nanocrystalline.

25 14. The piece according to any one of claims 1 to 13, characterized in that said least one nanocrystalline metal hydride is made of MgH_2 -5at. %.

30 15. The piece according to any one of claims 1 to 14, characterized in that the strip is of a shape selected from the group consisting of straight, stacked, folded, spiral, curved, twisted and cut shapes.

16. The piece according to any one of claims 1 to 15, characterized in that said piece is formed to have intrinsic electric characteristics allowing measurement of its hydrogen content.

5 17. The piece according to any one of claims 1 to 15, characterized in that said piece is formed to have intrinsic electric characteristics allowing desorption of hydrogen by circulation of an electrical current.

10 18. Use of a piece according to any one of claims 1 to 17 in a tank for storing and transporting hydrogen.

 19. Use of a piece according to any one of claims 1 to 17 in a battery of the Ni-MH type for stocking and transporting energy.

15 20. Use of a piece according to any one of claims 1 to 17 in equipment selected from the group consisting of purification, separation, compression, detection, cooling, heating, storing and generation of energy equipment.

20

**METAL HYDRIDE BANDS AND USE THEREOF
FOR STORING HYDROGEN**

5

BACKGROUND OF THE INVENTION

The present invention relates to pieces made of at least one metal hydride, which are in the form of strips obtained by rolling.

10 The invention also relates to the use of these pieces for the storage and transportation of hydrogen in hydrogen storage tanks.

The invention further relates to the use of the same pieces in batteries of the Ni-MH type and in equipment used for purification, separation, compression, detection, cooling, heating, storing and generation of energy.

15 In the present description and the appended claims, the expression "metal hydrides" is used to designate all [the] metals, alloys, composites and other materials known to absorb hydrogen in a reversible manner, whether these products are in a non-hydrogenated form (that is before absorption or after desorption of hydrogen) or in a hydrogenated form (that is after absorption and
20 before desorption of hydrogen). These hydrides can be of the type known to be [operative at] "high [temperatures] temperature". By way of examples of such hydrides, mention can be made to Na, Mg, Mg₂Ni, Li, Ti, Zr and Ca. These hydrides may also be of the type [operative at] "low or relatively low [temperatures] temperature". By way of examples of such hydrides, mention can
25 be made to MmNi₅, LaNi₅, CaNi₅, FeTi, Ti_{0.98} Zr_{0.02} V_{0.43} Fe_{0.09} Cr_{0.051} Mn_{1.5} and the Bogdanovic alloys.

BRIEF DESCRIPTION OF THE PRIOR ART

30 The use of metal hydrides for storing and transporting hydrogen is well known *per se*. The hydrogen which is so stored is usually used as a source of energy for hydrogen [operated] powered vehicles, batteries or machines or other

kind of equipments used for purification, separation, compression, detection, cooling and heating.

If all these potential applications are extremely interesting, there is still a practical problem of [transfer of] heat and mass (hydrogen), transfer which is associated to the use of metal hydrides having high hydrogen absorption and desorption kinetics. Indeed, absorption of hydrogen by a metal hydride is an exothermic phenomenon. Therefore, it is necessary to quickly evacuate the heat during the absorption, since otherwise the process will be slowed down. On the other hand, desorption is an endothermal phenomenon. It is therefore necessary to provide heat in a quick manner in order to extract hydrogen at a high rate from a metal hydride. When one wants to hydrogenate or dehydrogenate very quickly a material and thus, at the same time, to extract or supply heat, the thermal and diffusion [routes] paths must be reduced as much as possible, thereby making it necessary for the material to have at least one dimension which is small. Moreover, in order to obtain a high storage capacity per volume unit, it is necessary to consolidate and densify the hydride when the latter is in the form of a powder.

In order to tentatively overcome this problem, Japanese patent application No. JP-A-60/262.830 published on December 26, 1985 in the name of TOYOTA CENTER OF RESEARCH & DEVELOPMENT LAB suggests to give the shape of a thin sheet to the hydride. This sheet is obtained by "molding" the hydride around a metal mesh by means of a synthetic resin, such as a silicon resin. Such a "molding" process is quite difficult to carry out from an industrial standpoint. Moreover, as everybody knows, synthetic resins (polymers) cannot be exposed to high temperatures. Accordingly, the hydrides that can be used can only be low temperature metal hydrides (example 1 of this laid-open application refers to LaNi_5). It is impossible to use this technology with high temperature metal hydrides. Moreover, since synthetic resins do not conduct electricity and heat, it is necessary to incorporate heat pipes into the molded sheet in order to circulate a [heat transporting liquid] heating/coding fluid for heat exchange.

OBJECTS AND SUMMARY OF THE INVENTION

A first object of the present invention is to give to a metal hydride a form suitable to efficiently solve the practical problem [of transfers] of heat and
5 mass transfers mentioned hereinabove.

A second object of the invention is to achieve the requested forming of a metal hydride by using a technology that can easily be carried out in [high scale] mass production and at low cost [industrial productions].

A third object of the invention is to generalize the use of metal
10 hydrides that are so formed to all possible industrial applications for this type of product known to be capable of absorbing hydrogen in reversible manner.

In accordance with the invention, these various objects are achieved with at least one piece made of at least one metal hydride capable of absorbing hydrogen in a reversible manner, said piece [characterized in that it is] being in
15 the form of a thin and dense strip obtained by rolling of a powder of said at least one metal hydride and being characterized in that said strip is obtained at a temperature lower than 400°C.

This piece preferably has a thickness equal to or lower than 1 mm and made from a powder metal hydride(s) of nanocrystalline structure (cristallites of a
20 size lower than 100 nm). This piece can have any shape: straight, stacked, folded, spiral, curved, twisted or cut.

In accordance with a particularly preferred embodiment of the invention, this piece may contain a first additional component for supplying and/or evacuating heat. This first additional component may also act as a binder for the
25 powder of metal hydride(s).

The piece may also contain a second additional component acting as a binder for the powder of metal hydride(s).

The first and/or second additional components are preferably in the form of a powder additive. This additive may be a powder of Mg, Cu or Al.

Alternatively, the first and/or second additional components may be in the form of a tridimensional matrix preferably made of metal and having porous structure, that is rolled together with the powder of metal hydride(s).

The first and/or second additional components may also be in the form of a plate or tube in direct contact with the powder of metal hydride(s).

Preferably, the weight of the first and/or second additional components corresponds to a maximum of up to 50% of the weight of the whole piece. More preferably, this weight does not exceed 30% only of the weight of the whole piece.

Intrinsically, the piece according to the invention is capable of absorbing hydrogen in a reversible manner. It can be formed to have intrinsic electric characteristics allowing measurement of its hydrogen content. It can also be formed to have intrinsic electric characteristics allowing desorption of hydrogen by circulation of an electric current.

As indicated hereinabove, the piece made of metal hydride(s) according to the invention is manufactured by rolling. This process is particularly advantageous in that it allows the manufacture of very dense strips of a very small thickness which as a result, have the above-mentioned characteristics. This process also has the advantage of being perfectly adapted for industrial production on a large scale and at low cost.

Because of its property, the piece according to the invention is particularly well adapted for use as basic components in a tank for the storage and transportation of hydrogen. It can also be used in a battery of the type Ni-MH for the storage and transportation of energy. It can further be used in other equipment selected amongst the purification, separation, compression, detection, cooling, heating, storing and generation of energy [equipment].

The invention and its advantages will be better understood upon reading the following more detailed but non-restrictive description made in reference to the accompanying drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a schematic representation of the manufacture of a piece of metal hydride according to the invention, incorporating a tridimensional porous metal structure;

5 Figure 2 is a representation of a piece of metal hydride according to the invention, associated with, on the one hand, a metal plate used as a heat carrier (and possibly as a contact electrode) and, on the other hand, mattresses of porous fibers for the supply and removal of hydrogen and for volume compensation;

10 Figure 3 is a cross-sectional view of [a combination] an ensemble consisting of a piece according to the invention associated with, on the one hand, a metal plate acting as a heat carrier and, on the other hand, an insulating mattress, said figure giving a numerical simulation of the heat transfer in this [combination] ensemble during hydrogen absorption;

15 Figure 4 is a curve illustrating the variation of resistivity of a consolidated piece of a metal hydride of the formula $\text{MgH}_2\text{-5at.\%V}$ during hydrogen desorption and the variation of differential pressure of the hydrogen titration apparatus, as a function of the temperature;

20 Figure 5 is a schematic cross-sectional view of a tank for storing hydrogen, comprising a stack of pieces according to the invention;

 Figure 6 is an open perspective view of the tank shown in cross-section in Figure 5; and

 Figures 7a to 7b are perspective views showing in a schematic manner, different shapes that may have the pieces according to the invention.

25

DETAILED DESCRIPTION OF THE INVENTION

As previously indicated, the invention essentially relates to a piece made of one or more metal hydrides that are capable of absorbing hydrogen in a reversible manner. The piece is in the form of a thin and dense strip having a
30 thickness preferably equal to or lower than 1 mm and a porosity preferably lower

than 20% which is obtained by rolling a powder of the selected hydride(s) with or without additional component(s). This strip has a typical width which is preferably lower than 15 cm and a length that can be of several meters.

By way of non-restrictive examples of hydrides that can be used for the manufacture of the pieces, mention can be made to those of nanocrystalline structures disclosed in U.S. patent Nos. [5,763,360] 5,763,363; 5,837,030; 5,872,074 and 5,906,792 either in the name of the Applicant alone or in the name of the Applicant and the MCGILL UNIVERSITY. Even more particularly mention can be made to Mg.-based hydrides.

The key feature of the invention lies in the use of the well known method of "rolling" for giving to the powder of hydrides the form of thin and dense strips and thus for allowing optimization of the transfers of mass (hydrogen) and heat involved during the process of absorption/desorption of hydrogen.

In order to [so] obtain by rolling strips having a high structural integrity, additional components such as binders can be added as additives to the hydrides. By way of examples of such additives, reference can be made to Mg, Al, V, Cu, Li, Fe, Nb, C, graphite, Co, Ni, Mn, Cr, Ti, Zr and FeTi. These additives are preferably in the form of a powder that can be mixed with the powder of hydride(s) and rolled together with it.

As an additional component, it may also be suggested to use a tridimensional structure preferably porous and made of metal, such as a metal foam, to insert the metal hydride powder into the pores of the structure and to roll the whole to obtain a thin strip. The metal structure that is so incorporated to the piece then permits to more quickly remove heat from the strip during hydrogen absorption.

Figure 1 is illustrative of this process of manufacture. As is shown, the tridimensional metal structure 3 is introduced with a carrier plate 5 preferably made of a heat conductive material, between the rollers 7 of a rolling machine. Just before such an introduction, the powder of metal hydride 9 stored into a

tank 11 is poured into the structure 3 to fill up its pores. After rolling, the so obtained piece in the form of a thin and dense strip 1 can then be combined with other elements such as an electrical contact grid 13 and mattresses 15 of micro glass-fibers in order to obtain an operating unit ready to be incorporated into a tank for the storage of hydrogen.

In use, the rolling step can be carried out at ambient temperature or at elevated temperature (lower than 400°C). The rolls 7 can be heated and their relative position can be adjusted to obtain a requested thickness and density.

As can be understood, the above-mentioned additives are used and rolled with the powder of metal hydride essentially in order to give structural integrity to the strip and to act as a heat conductor. Accordingly, it is important to preferably use additives in the form of metal powders or metal fibers, which are capable of acting not only as binders but also as heat conductors.

Together with these additives or as substituents for them, use can also be made of the tridimensional metal structure, or of the carrier plate or grid as already referred to hereinabove, or also of a metal tube into which the hydride powder is inserted before the rolling step. These components should also be made of a heat conductive material and be preferably provided with holes to let flow the hydrogen to be absorbed or desorbed.

In order to promote the desorption of hydrogen, use can be made of the intrinsic electric characteristics of the pieces in the form of strips obtained by rolling. Thus, for example, it is known that magnesium hydride MgH_2 is an insulating material. When the content of metal additives in a form of a powder or in a form of porous structure exceeds the percolation point, a strip made from this hydride becomes conductive. Moreover, during the desorption, MgH_2 switches from an insulating state to a conductive state. Thus, use can be made of these characteristics to induce desorption of hydrogen from the strip by passing an electric current therein. The change of resistivity of the pieces as a function of the content of hydrogen contained in them may also be used as an indicator of the amount of hydrogen within the tank.

Figure 4 illustrates as an example, the change of resistivity of a consolidated piece made of MgH_2 and the change of differential pressure of the hydrogen titration apparatus used for measuring the amount of hydrogen within this piece as a function of the temperature during a desorption step. This figure shows that, during desorption, magnesium hydride (MgH_2) which is an insulating material, becomes conductive (Mg). This figure can be associated to the table given below which indicates the value of the electric resistance of 1 mm thick rolled pieces made from the same nanocrystalline metal hydride of formula MgH_2 -5at%V, to which were added variable amounts of a binder consisting of Mg of granulometry ranging from $105\mu\text{m}$ to $850\mu\text{m}$ (the internal resistance of the measurement apparatus was 2000Ω).

Amount of binder (Mg) added (expressed in weight)	0	10%	15%	20%	30%
Measured resistance (Ω)	4×10^{10}	1×10^{10}	3000	<2000	<2000

It will be noticed that when the additional component acting as a binder is in the form of a powder that is mixed with and integrated to the hydride powder, it is of the highest importance that this component be well distributed and occupy all the volume of the piece, as is the case of a tri-dimensional metal structure when it is inserted into the hydride powder during rolling (see Figure 1). As a matter of fact, one and/or the other of these two solutions are very preferable to a mere pressing of the hydride powder onto a substrate without any additives.

The concept of using pieces in the form of rolled strips of metal hydrides prepared as disclosed hereinabove, is sufficiently flexible to be adapted to a multitude of applications. These strips may be of any shape. They may be flat and rectilinear strips, cut into straight sections (see figures 5 and 6). Alternatively, the strips may be folded and stacked upon themselves (figure 7a),

rolled in a spiral (figure 7b), curved (figure 7C) or cut, sectioned and reassembled at an angle (figure 7d).

Figures 5 and 6 illustrate a possible use of several pieces 1 in the form of straight strips into a tank 21 for the storage and transportation of hydrogen. The tank 21 comprises an external wall 23, an internal wall 25 fixed to the external wall via a corrugated sleeve 27 acting as a mechanical reinforcement and a thermal breaking. Thanks to its structure, the sleeve also defines a plurality of conducts 29 for a heat carrying fluid (water, oil or any other liquid) which is brought at 31 and used for cooling during the absorption or for heating during the desorption of hydrogen. The hydrogen circulates via a pipe 33 opening into the container which is defined by the internal wall 25 and in which the pieces 1 are stacked. Contact plates 5 perforated with holes, act as heat carriers and are connected to the internal wall to act together with it as a heat exchanger. Mattresses of glass microfibers having a porosity of, for example, 90% with pores of, for example, 0,3 microns, are positioned between the pieces as shown in figure 2, to ensure the supply and removal of hydrogen and to absorb volume dilatation or contraction.

The tank 21 disclosed hereinabove, has numerous advantages. It is safe and of easy operation. It is easy to fill and of a great capacity. It allows liberation of hydrogen upon request, and such in a very efficient and fast manner thanks to the structure of the pieces 1 and their spacing. The absorption/desorption kinetics are very fast and the storage capacity is very high. The number of cycles of absorption/desorption is also high.

Such makes the tank 21 useful in the field of transportation (cars, buses, tractors, trucks, recreation vehicles, planes, boats, trains, military transports, etc...) or for the manufacturers/users of fuel cells, the producers of decentralized energy, the users of hydrogen of very high purity and the users of filters for hydrogen.

So far, in all the applications involving hydrogen especially as a source of energy for fuel cells or internal combustion engines or as sources for the

production of hydrogen in reformers or electrocatalic machines, the storage of hydrogen was done either under very high pressure with the safety problem that such generates (pressure $\gg 300$ psi), or in liquid form, with the problems [to] of the storage at very low temperatures ($< -253^{\circ}\text{C}$). Another solution is the use of

5 conventional hydrides in the form of a non compacted powder or of massive pieces with the problems that such generate, in particular, a very low absorption/desorption kinetics and a low storage capacity.

The use of [a] rolling as proposed according to the invention for the manufacture of thin and dense strips of hydrides, permits to optimize the

10 absorption/desorption kinetics in storage tanks and to maximize the heat transfer. The use of the intrinsic electric properties of the consolidated strips close to the percolation point, permits to measure the content of hydrogen and promote the desorption of hydrogen. Because of their structure, the so-obtained pieces permit to optimize mass and heat transfer within the tanks of metal hydrides. [The] It

15 provides large hydrogen absorption/desorption kinetics, [provide a] fast mass [in a] and heat transfer and high number of cycles of absorption/desorption. Also because of their structure, the consolidated pieces are safe as compared to a powder of non-binded hydride which presents a [given degree] certain level of pyrophoricity.

20 The technology disclosed hereinabove and the pieces of hydride(s) in the form of strips according to the invention may also be used efficiently for applications in batteries of the Ni-MH type (nickel-metal hydride).

The field of application of the hydride strips and the hydrogen storage tanks using these pieces in the form of rolled hydride strips according to the

25 invention is therefore very wide. This technology is particularly well adapted for use with nanocrystalline hydrides since these materials have very high hydrogen absorption and desorption kinetics. It is particularly well adapted for the Mg-containing hydrides (Mg, Mg_2Ni and the related materials), Li, Na, Ti, Zr and Ca, which are known to be "stable" hydrides operating at a high temperature and for

30 which the problem of heat exchange is important. This technology may also be

5 nanotubes, etc...

CLAIMS

1. A piece made of at least one metal hydride capable of absorbing
5 hydrogen in a reversible manner, said piece being in the form of a thin and dense
strip obtained by rolling a powder of said at least one metal hydride, characterized
in that said strip is obtained at a temperature lower than 400°C.

2. The piece according to claim 1, characterized in that it comprises a
10 first additional component for supplying and/or evacuating heat.

3. The piece according to claim 2, characterized in that the first
additional component is also acting as a binder for the powder of said at least one
metal hydride.

15

4. The piece according to claim 1 or 2, characterized in that it
comprises a second additional component acting as a binder for the powder of
said at least one metal hydride.

5. The piece according to claim 2, 3, or 4, characterized in that the
20 first and/or second additional component is in the form of a powder additive.

6. The piece according to claim 5, characterized in that the powder
additive consists of Mg.

25

7. The piece according to claim 2, 3 or 4, characterized in that the first
and/or second additional component is in the form of a tridimensional matrix that
is rolled together with the powder of said at least one metal hydride.

30

8. The piece according to claim 7, characterized in that the matrix is made of metal and has a porous structure.

5 9. The piece according to claim 2, 3 or 4, characterized in that the first and/or second additional component is in the form of a plate in direct contact with the powder of said at least one metal hydride or of a tube containing said at least one metal hydride or of a rolled tube containing said powder of said at least one metal hydride.

10 10. The piece according to any one of claims 2 to 9, wherein the first and/or second additional components have a weight amounting to up to 50% of the weight of the whole piece.

15 11. The piece according to claim 10, characterized in that the first and/or second additional components have a weight amounting to up to 30% of the weight of the whole piece.

20 12. The piece according to any one of claims 1 to 11, characterized in that the strip has a thickness equal to or lower than 1 mm.

13. The piece according to any one of claims 1 to 12, characterized in that the powder of said at least one metal hydride is nanocrystalline.

25 14. The piece according to any one of claims 1 to 13, characterized in that said least one nanocrystalline metal hydride is made of MgH_2 -5at. %.

30 15. The piece according to any one of claims 1 to 14, characterized in that the strip is of a shape selected from the group consisting of straight, stacked, folded, spiral, curved, twisted and cut shapes.

16. The piece according to any one of claims 1 to 15, characterized in that said piece is formed to have intrinsic electric characteristics allowing measurement of its hydrogen content.

5 17. The piece according to any one of claims 1 to 15, characterized in that said piece is formed to have intrinsic electric characteristics allowing desorption of hydrogen by circulation of an electrical current.

10 18. Use of a piece according to any one of claims 1 to 17 in a tank for storing and transporting hydrogen.

 19. Use of a piece according to any one of claims 1 to 17 in a battery of the Ni-MH type for stocking and transporting energy.

15 20. Use of a piece according to any one of claims 1 to 17 in equipment selected from the group consisting of purification, separation, compression, detection, cooling, heating, storing and generation of energy equipment.

EXPRESS MAIL CERTIFICATE

Date 2/15/02 Label No. EV039137720US

I hereby certify that, on the date indicated above, this paper or fee was deposited with the U.S. Postal Service & that it was addressed for delivery to the Assistant Commissioner for Patents, Washington, DC 20231 by "Express Mail Post Office to Addressee" service.

PLEASE CHARGE ANY DEFICIENCY UP TO \$300.00 OR CREDIT ANY EXCESS IN THE FEES DUE WITH THIS DOCUMENT TO OUR DEPOSIT ACCOUNT NO. 04-0100

Name (Print)

Signature

Customer No.:



07278

PATENT TRADEMARK OFFICE

Docket No: 6670/OK215

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re Application of: Robert SCHULZ et al.

Serial No: T/B/A (National Phase of PCT/CA00/00865)

Filed: Concurrently Herewith

For: METAL HYDRIDE BANDS AND USE THEREOF FOR STORING HYDROGEN

Hon. Commissioner for Patents
Washington, DC 20231

**TRANSMITTAL OF ENGLISH TRANSLATION OF
INTERNATIONAL APPLICATION (35 U.S.C. §371 (c) (3))**

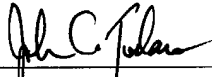
Sir:

Attached is an English translation of International Application No. PCT/CA00/00865 as originally filed, of which this application is a national

phase entry under 35 U.S.C. § 371.

Respectfully submitted,

Dated: February 15, 2002



John C. Todaro
Reg. No. 36,036
Attorney for Applicant(s)

DARBY & DARBY P.C.
805 Third Avenue
New York, NY 10022
212-527-7700

::ODMA\WORLD\OX\M:\6670\OK215\LWJ7465 WPD

Serial No. T/B/A
(National Phase of PCT/CA00/00865)

Docket No. 6670/OK215

METAL HYDRIDE BANDS AND USE THEREOF FOR STORING HYDROGEN

5

BACKGROUND OF THE INVENTION

The present invention relates to pieces made of at least one metal hydride, which are in the form of strips obtained by rolling.

10 The invention also relates to the use of these pieces for the storage and transportation of hydrogen in hydrogen storage tanks.

The invention further relates to the use of the same pieces in batteries of the Ni-MH type and in equipment used for purification, separation, compression, detection, cooling, heating, storing and generation of energy.

15 In the present description and the appended claims, the expression "metal hydrides" is used to designate all the metals, alloys, composites and other materials known to absorb hydrogen in a reversible manner, whether these products are in a non-hydrogenated form (that is before absorption or after desorption of hydrogen) or in a hydrogenated form (that is after absorption and before desorption of hydrogen). These hydrides can be of the type known to be
20 operative at "high temperatures". By way of examples of such hydrides, mention can be made to Mg, Mg₂Ni, Li, Ti, Zr and Ca. These hydrides may also be of the type operative at "low or relatively low temperatures". By way of examples of such hydrides, mention can be made to MmNi₅, LaNi₅, CaNi₅, FeTi, Ti_{0.98}Zr_{0.02}V_{0.43}Fe_{0.09}Cr_{0.051}Mn_{1.5} and the Bogdanovic alloys.

25

BRIEF DESCRIPTION OF THE PRIOR ART

The use of metal hydrides for storing and transporting hydrogen is well known *per se*. The hydrogen which is so stored is usually used as a source of energy for hydrogen-operated vehicles, batteries or machines or other kind of
30 equipments used for purification, separation, compression, detection, cooling and heating.

If all these potential applications are extremely interesting, there is still a practical problem of transfer of heat and mass (hydrogen), which is associated to the use of metal hydrides having high hydrogen absorption and desorption kinetics. Indeed, absorption of hydrogen by a metal hydride is an exothermic phenomenon. Therefore, it is necessary to quickly evacuate the heat during the absorption, since otherwise the process will be slowed down. On the other hand, desorption is an endothermal phenomenon. It is therefore necessary to provide heat in a quick manner in order to extract hydrogen at a high rate from a metal hydride. When one wants to hydrogenate or dehydrogenate very quickly a material and thus, at the same time, to extract or supply heat, the thermal and diffusion routes must be reduced as much as possible, thereby making it necessary for the material to have at least one dimension which is small. Moreover, in order to obtain a high storage capacity per volume unit, it is necessary to consolidate and densify the hydride when the latter is in the form of a powder.

In order to tentatively overcome this problem, Japanese patent application No. JP-A-60/262.830 published on December 26, 1985 in the name of TOYOTA CENTER OF RESEARCH & DEVELOPMENT LAB suggests to give the shape of a thin sheet to the hydride. This sheet is obtained by "molding" the hydride around a metal mesh by means of a synthetic resin, such as a silicon resin. Such a "molding" process is quite difficult to carry out from an industrial standpoint. Moreover, as everybody knows, synthetic resins (polymers) cannot be exposed to high temperatures. Accordingly, the hydrides that can be used can only be low temperature metal hydrides (example 1 of this laid-open application refers to LaNi_5). It is impossible to use this technology with high temperature metal hydrides. Moreover, since synthetic resins do not conduct electricity and heat, it is necessary to incorporate pipes into the molded sheet in order to circulate a heat-transporting liquid for heat exchange.

OBJECTS AND SUMMARY OF THE INVENTION

A first object of the present invention is to give to a metal hydride a form suitable to efficiently solve the practical problem of transfers of heat and mass mentioned hereinabove.

5 A second object of the invention is to achieve the requested forming of a metal hydride by using a technology that can easily be carried out in high scale and low cost industrial productions.

10 A third object of the invention is to generalize the use of metal hydrides that are so formed to all possible industrial applications for this type of product known to be capable of absorbing hydrogen in reversible manner.

In accordance with the invention, these various objects are achieved with at least one piece made of at least one metal hydride capable of absorbing hydrogen in a reversible manner, characterized in that it is in the form of a thin and dense strip obtained by rolling of a powder of said at least one metal hydride.

15 This piece preferably has a thickness equal to or lower than 1 mm and made from a powder metal hydride(s) of nanocrystalline structure (crystallites of a size lower than 100 nm). This piece can have any shape: straight, stacked, folded, spiral, curved, twisted or cut.

20 In accordance with a particularly preferred embodiment of the invention, this piece may contain a first additional component for supplying and/or evacuating heat. This first additional component may also act as a binder for the powder of metal hydride(s).

The piece may also contain a second additional component acting as a binder for the powder of metal hydride(s).

25 The first and/or second additional components are preferably in the form of a powder additive. This additive may be a powder of Mg, Cu or Al.

Alternatively, the first and/or second additional components may be in the form of a tridimensional matrix preferably made of metal and having porous structure, that is rolled together with the powder of metal hydride(s).

The first and/or second additional components may also be in the form of a plate or tube in direct contact with the powder of metal hydride(s).

Preferably, the weight of the first and/or second additional components corresponds to a maximum of up to 50% of the weight of the whole piece. More preferably, this weight does not exceed 30% only of the weight of the whole piece.

Intrinsically, the piece according to the invention is capable of absorbing hydrogen in a reversible manner. It can be formed to have intrinsic electric characteristics allowing measurement of its hydrogen content. It can also be formed to have intrinsic electric characteristics allowing desorption of hydrogen by circulation of an electric current.

As indicated hereinabove, the piece made of metal hydride(s) according to the invention is manufactured by rolling. This process is particularly advantageous in that it allows the manufacture of very dense strips of a very small thickness which as a result, have the above-mentioned characteristics. This process also has the advantage of being perfectly adapted for industrial production on a large scale and at low cost.

Because of its property, the piece according to the invention is particularly well adapted for use as basic components in a tank for the storage and transportation of hydrogen. It can also be used in a battery of the type Ni-MH for the storage and transportation of energy. It can further be used in other equipment selected amongst the purification, separation, compression, detection, cooling, heating, storing and generation of energy equipment.

The invention and its advantages will be better understood upon reading the following more detailed but non-restrictive description made in reference to the accompanying drawings.

BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a schematic representation of the manufacture of a piece of metal hydride according to the invention, incorporating a tridimensional porous

metal structure;

Figure 2 is a representation of a piece of metal hydride according to the invention, associated with, on the one hand, a metal plate used as a heat carrier (and possibly as a contact electrode) and, on the other hand, mattresses of porous fibers for the supply and removal of hydrogen and for volume compensation;

Figure 3 is a cross-sectional view of a combination consisting of a piece according to the invention associated with, on the one hand, a metal plate acting as a heat carrier and, on the other hand, an insulating mattress, said figure giving a numerical simulation of the heat transfer in this combination during hydrogen absorption;

Figure 4 is a curve illustrating the variation of resistivity of a consolidated piece of a metal hydride of the formula $MgH_2-5at.\%V$ during hydrogen desorption and the variation of differential pressure of the hydrogen titration apparatus, as a function of the temperature;

Figure 5 is a schematic cross-sectional view of a tank for storing hydrogen, comprising a stack of pieces according to the invention;

Figure 6 is an open perspective view of the tank shown in cross-section in Figure 5; and

Figures 7a to 7b are perspective views showing in a schematic manner, different shapes that may have the pieces according to the invention.

DETAILED DESCRIPTION OF THE INVENTION

As previously indicated, the invention essentially relates to a piece made of one or more metal hydrides that are capable of absorbing hydrogen in a reversible manner. The piece is in the form of a thin and dense strip having a thickness preferably equal to or lower than 1 mm and a porosity preferably lower than 20% which is obtained by rolling a powder of the selected hydride(s) with or without additional component(s). This strip has a typical width which is preferably lower than 15 cm and a length that can be of several meters.

By way of non-restrictive examples of hydrides that can be used for the manufacture of the pieces, mention can be made to those of nanocrystalline structures disclosed in U.S. patent Nos. 5,763,360; 5,837,030; 5,872,074 and 5,906,792 either in the name of the Applicant alone or in the name of the Applicant and the MCGILL UNIVERSITY. Even more particularly mention can be made to Mg.-based hydrides.

The key feature of the invention lies in the use of the well known method of "rolling" for giving to the powder of hydrides the form of thin and dense strips and thus for allowing optimization of the transfers of mass (hydrogen) and heat involved during the process of absorption/desorption of hydrogen.

In order to so obtain by rolling strips having a high structural integrity, additional components such as binders can be added as additives to the hydrides. By way of examples of such additives, reference can be made to Mg, Al, V, Cu, Li, Fe, Nb, C, graphite, Co, Ni, Mn, Cr, Ti, Zr and FeTi. These additives are preferably in the form of a powder that can be mixed with the powder of hydride(s) and rolled together with it.

As an additional component, it may also be suggested to use a tridimensional structure preferably porous and made of metal, such as a metal foam, to insert the metal hydride powder into the pores of the structure and to roll the whole to obtain a thin strip. The metal structure that is so incorporated to the piece then permits to more quickly remove heat from the strip during hydrogen absorption.

Figure 1 is illustrative of this process of manufacture. As is shown, the tridimensional metal structure 3 is introduced with a carrier plate 5 preferably made of a heat conductive material, between the rollers 7 of a rolling machine. Just before such an introduction, the powder of metal hydride 9 stored into a tank 11 is poured into the structure 3 to fill up its pores. After rolling, the so obtained piece in the form of a thin and dense strip 1 can then be combined with other elements such as an electrical contact grid 13 and mattresses 15 of micro

glass-fibers in order to obtain an operating unit ready to be incorporated into a tank for the storage of hydrogen.

In use, the rolling step can be carried out at ambient temperature or at elevated temperature (lower than 400°C). The rolls 7 can be heated and their relative position can be adjusted to obtain a requested thickness and density.

As can be understood, the above-mentioned additives are used and rolled with the powder of metal hydride essentially in order to give structural integrity to the strip and to act as a heat conductor. Accordingly, it is important to preferably use additives in the form of metal powders or metal fibers, which are capable of acting not only as binders but also as heat conductors.

Together with these additives or as substituents for them, use can also be made of the tridimensional metal structure, or of the carrier plate or grid as already referred to hereinabove, or also of a metal tube into which the hydride powder is inserted before the rolling step. These components should also be made of a heat conductive material and be preferably provided with holes to let flow the hydrogen to be absorbed or desorbed.

In order to promote the desorption of hydrogen, use can be made of the intrinsic electric characteristics of the pieces in the form of strips obtained by rolling. Thus, for example, it is known that magnesium hydride MgH_2 is an insulating material. When the content of metal additives in a form of a powder or in a form of porous structure exceeds the percolation point, a strip made from this hydride becomes conductive. Moreover, during the desorption, MgH_2 switches from an insulating state to a conductive state. Thus, use can be made of these characteristics to induce desorption of hydrogen from the strip by passing an electric current therein. The change of resistivity of the pieces as a function of the content of hydrogen contained in them may also be used as an indicator of the amount of hydrogen within the tank.

Figure 4 illustrates as an example, the change of resistivity of a consolidated piece made of MgH_2 and the change of differential pressure of the hydrogen titration apparatus used for measuring the amount of hydrogen within

this piece as a function of the temperature during a desorption step. This figure shows that, during desorption, magnesium hydride (MgH_2) which is an insulating material, becomes conductive (Mg). This figure can be associated to the table given below which indicates the value of the electric resistance of 1 mm thick rolled pieces made from the same nanocrystalline metal hydride of formula MgH_2 -5at%V, to which were added variable amounts of a binder consisting of Mg of granulometry ranging from $105\mu\text{m}$ to $850\mu\text{m}$ (the internal resistance of the measurement apparatus was 2000Ω).

Amount of binder (Mg) added (expressed in weight)	0	10%	15%	20%	30%
Measured resistance (Ω)	4×10^{10}	1×10^{10}	3000	<2000	<2000

10

It will be noticed that when the additional component acting as a binder is in the form of a powder that is mixed with and integrated to the hydride powder, it is of the highest importance that this component be well distributed and occupy all the volume of the piece, as is the case of a tri-dimensional metal structure when it is inserted into the hydride powder during rolling (see Figure 1). As a matter of fact, one and/or the other of these two solutions are very preferable to a mere pressing of the hydride powder onto a substrate without any additives.

The concept of using pieces in the form of rolled strips of metal hydrides prepared as disclosed hereinabove, is sufficiently flexible to be adapted to a multitude of applications. These strips may be of any shape. They may be flat and rectilinear strips, cut into straight sections (see figures 5 and 6). Alternatively, the strips may be folded and stacked upon themselves (figure 7a), rolled in a spiral (figure 7b), curved (figure 7c) or cut, sectioned and reassembled at an angle (figure 7d).

25

Figures 5 and 6 illustrate a possible use of several pieces 1 in the form of straight strips into a tank 21 for the storage and transportation of hydrogen. The tank 21 comprises an external wall 23, an internal wall 25 fixed to the external wall via a corrugated sleeve 27 acting as a mechanical reinforcement and a thermal breaking. Thanks to its structure, the sleeve also defines a plurality of conducts 29 for a heat carrying fluid (water, oil or any other liquid) which is brought at 31 and used for cooling during the absorption or for heating during the desorption of hydrogen. The hydrogen circulates via a pipe 33 opening into the container which is defined by the internal wall 25 and in which the pieces 1 are stacked. Contact plates 5 perforated with holes, act as heat carriers and are connected to the internal wall to act together with it as a heat exchanger. Mattresses of glass microfibers having a porosity of, for example, 90% with pores of, for example, 0,3 microns, are positioned between the pieces as shown in figure 2, to ensure the supply and removal of hydrogen and to absorb volume dilatation or contraction.

The tank 21 disclosed hereinabove, has numerous advantages. It is safe and of easy operation. It is easy to fill and of a great capacity. It allows liberation of hydrogen upon request, and such in a very efficient and fast manner thanks to the structure of the pieces 1 and their spacing. The absorption/desorption kinetics are very fast and the storage capacity is very high. The number of cycles of absorption/desorption is also high.

Such makes the tank 21 useful in the field of transportation (cars, buses, tractors, trucks, recreation vehicles, planes, boats, trains, military transports, etc...) or for the manufacturers/users of fuel cells, the producers of decentralized energy, the users of hydrogen of very high purity and the users of filters for hydrogen.

So far, in all the applications involving hydrogen especially as a source of energy for fuel cells or internal combustion engines or as sources for the production of hydrogen in reformers or electrocatalic machines, the storage of hydrogen was done either under very high pressure with the safety problem that

such generates (pressure $> > 300$ psi), or in liquid form, with the problems to the storage at very low temperatures ($< -253^{\circ}\text{C}$). Another solution is the use of conventional hydrides in the form of a non compacted powder or of massive pieces with the problems that such generate, in particular, a very low
5 absorption/desorption kinetics and a low storage capacity.

The use of a rolling as proposed according to the invention for the manufacture of thin and dense strips of hydrides, permits to optimize the absorption/desorption kinetics in storage tanks and to maximize the heat transfer. The use of the intrinsic electric properties of the consolidated strips close to the
10 percolation point, permits to measure the content of hydrogen and promote the desorption of hydrogen. Because of their structure, the so-obtained pieces permit to optimize mass and heat transfer within the tanks of metal hydrides. The hydrogen absorption/desorption kinetics provide a fast mass and heat transfer in a high number of cycles of absorption/desorption. Also because of their structure,
15 the consolidated pieces are safe as compared to a powder of non-binded hydride which presents a given degree of pyrophoricity.

The technology disclosed hereinabove and the pieces of hydride(s) in the form of strips according to the invention may also be used efficiently for applications in batteries of the Ni-MH type (nickel-metal hydride).

20 The field of application of the hydride strips and the hydrogen storage tanks using these pieces in the form of rolled hydride strips according to the invention is therefore very wide. This technology is particularly well adapted for use with nanocrystalline hydrides since these materials have very high hydrogen absorption and desorption kinetics. It is particularly well adapted for the Mg-
25 containing hydrides (Mg, Mg_2Ni and the related materials), Li, Na, Ti, Zr and Ca, which are known to be "stable" hydrides operating at a high temperature and for which the problem of heat exchange is important. This technology may also be used with low temperature hydrides such as MmNi_5 , LaNi_5 , CaNi_5 , FeTi , $\text{Ti}_{0.98}$, $\text{Zr}_{0.02}$, $\text{V}_{0.43}$, $\text{Fe}_{0.09}$, $\text{Cr}_{0.05}$, $\text{Mn}_{1.5}$, of Bogdanovic alloys, etc., that is, in summary,

with all hydrides of the AB_5 , AB_2 , AB , A_2B , types, solid solutions, nanocrystalline and amorphous alloys, complex hydrides and even carbon, nanotubes, etc...

CLAIMS

1. A piece made of at least one metal hydride capable of absorbing
5 hydrogen in a reversible manner, characterized in that it is in the form of a thin
and dense strip obtained by rolling a powder of said at least one metal hydride.

2. The piece according to claim 1, characterized in that it comprises a
first additional component for supplying and/or evacuating heat.
10

3. The piece according to claim 2, characterized in that the first
additional component is also acting as a binder for the powder of said at least one
metal hydride.

4. The piece according to claim 1 or 2, characterized in that it
15 comprises a second additional component acting as a binder for the powder of
said at least one metal hydride.

5. The piece according to claim 2, 3, or 4, characterized in that the
20 first and/or second additional component is in the form of a powder additive.

6. The piece according to claim 5, characterized in that the powder
additive consists of Mg.

7. The piece according to claim 2, 3 or 4, characterized in that the first
25 and/or second additional component is in the form of a tridimensional matrix that
is rolled together with the powder of said at least one metal hydride.

8. The piece according to claim 7, characterized in that the matrix is
30 made of metal and has a porous structure.

5 9. The piece according to claim 2, 3 or 4, characterized in that the first and/or second additional component is in the form of a plate in direct contact with the powder of said at least one metal hydride or of a tube containing said at least one metal hydride or of a rolled tube containing said powder of said at least one metal hydride.

10 10. The piece according to any one of claims 2 to 9, wherein the first and/or second additional components have a weight amounting to up to 50% of the weight of the whole piece.

15 11. The piece according to claim 10, characterized in that the first and/or second additional components have a weight amounting to up to 30% of the weight of the whole piece.

20 12. The piece according to any one of claims 1 to 11, characterized in that the strip has a thickness equal to or lower than 1 mm.

13. The piece according to any one of claims 1 to 12, characterized in that the powder of said at least one metal hydride is nanocrystalline.

25 14. The piece according to any one of claims 1 to 13, characterized in that said least one nanocrystalline metal hydride is made of MgH_2 -5at. %.

30 15. The piece according to any one of claims 1 to 14, characterized in that the strip is of a shape selected from the group consisting of straight, stacked, folded, spiral, curved, twisted and cut shapes.

16. The piece according to any one of claims 1 to 15, characterized in that said piece is formed to have intrinsic electric characteristics allowing measurement of its hydrogen content.

5 17. The piece according to any one of claims 1 to 15, characterized in that said piece is formed to have intrinsic electric characteristics allowing desorption of hydrogen by circulation of an electrical current.

10 18. Use of a piece according to any one of claims 1 to 17 in a tank for storing and transporting hydrogen.

19. Use of a piece according to any one of claims 1 to 17 in a battery of the Ni-MH type for stocking and transporting energy.

15 20. Use of a piece according to any one of claims 1 to 17 in equipment selected from the group consisting of purification, separation, compression, detection, cooling, heating, storing and generation of energy equipment.

(12) DEMANDE INTERNATIONALE PUBLIÉE EN VERTU DU TRAITÉ DE COOPÉRATION
EN MATIÈRE DE BREVETS (PCT)(19) Organisation Mondiale de la Propriété
Intellectuelle
Bureau international(43) Date de la publication internationale
22 février 2001 (22.02.2001)

PCT

(10) Numéro de publication internationale
WO 01/13445 A1(51) Classification internationale des brevets⁷: H01M 4/24,
8/06, C01B 3/00, 6/24, B22F 3/11(21) Numéro de la demande internationale:
PCT/CA00/00865

(22) Date de dépôt international: 21 juillet 2000 (21.07.2000)

(25) Langue de dépôt: français

(26) Langue de publication: français

(30) Données relatives à la priorité:
2,280,434 18 août 1999 (18.08.1999) CA(71) Déposant (pour tous les États désignés sauf US): HY-
DRO-QUEBEC [CA/CA]; 75 René-Lévesque Ouest,
Montréal, Québec H2Z 1A4 (CA).

(72) Inventeurs; et

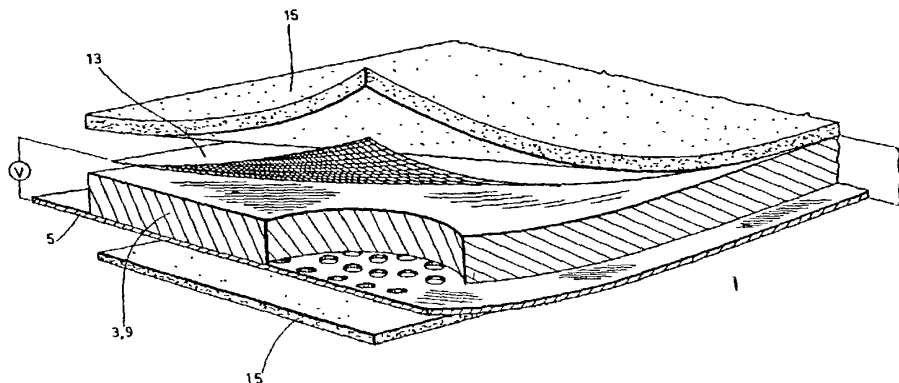
(75) Inventeurs/Déposants (pour US seulement): SCHULZ,
Robert [CA/CA]; 68 Bel Horizon, Sainte-Julie, Québec
J0L 2S0 (CA). BOILY, Sabin [CA/CA]; 1676 Labonté,
Chambly, Québec J3L 5M6 (CA). DUBUC, René
[CA/CA]; 380 De Normandie, Sainte-Julie, Québec J3E
1E8 (CA). BLOUIN, Marco [CA/CA]; 4601, 18e Avenue,
app. 8, Montréal, Québec H1X 2N6 (CA). LALANDE,
Guy [CA/CA]; 358 St-Joseph Est, Montréal, Québec H2T
1J4 (CA).(74) Mandataire: ROBIC; 55 St-Jacques, Montréal, Québec
H2Y 3X2 (CA).

(81) États désignés (national): BR, CN, JP, KR, MX, US.

[Suite sur la page suivante]

(54) Title: METAL HYDRIDE BANDS AND USE THEREOF FOR STORING HYDROGEN

(54) Titre: RUBANS D'HYDRURES METALLIQUES ET LEUR USAGE POUR LE STOCKAGE DE L'HYDROGENE



(57) Abstract: The invention concerns a piece based on one or several metal hydrides capable of reversibly absorbing hydrogen. Said piece is in the form of a thin and dense band, having a thickness preferably not more than 1 mm and porosity preferably less than 20 %. The piece is obtained by rolling a powder of selected hydride(s), with or without additional component(s), such as binders or heat-transfer elements. Said piece can easily be produced on an industrial scale. By its very nature, it is particularly adapted for use as a base element in a tank for storing and transporting hydrogen. It can also be used in a Ni-MH type battery for storing and transporting energy.

(57) Abrégé: L'invention vise une pièce à base d'un ou plusieurs hydrures métalliques capables d'absorber l'hydrogène de façon réversible. Cette pièce se présente sous la forme d'un ruban mince et dense, d'épaisseur de préférence égale ou inférieure à 1 mm et de porosité de préférence inférieure à 20%. La pièce est obtenue par laminage d'une poudre du ou des hydrures choisis, avec ou sans composante(s) additionnelle(s), telles que des liants ou éléments caloporteurs. Cette pièce peut facilement être produite à une échelle industrielle. De par sa nature, elle est particulièrement bien adaptée pour être utilisée comme élément de base dans un réservoir pour le stockage et le transport de l'hydrogène. Elle peut aussi être utilisée dans une batterie du type Ni-MH pour le stockage et le transport d'énergie.



10059035 . 052202US

COMBINED DECLARATION FOR PATENT APPLICATION AND POWER OF ATTORNEY

(Includes Reference to PCT International Applications)

ATTORNEY DOCKET NUMBER
6670/OK215US0

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed for and which a patent is sought on the invention entitled:

METAL HYDRIDE BANDS AND USE THEREOF FOR STORING HYDROGEN

the specification of which (check only one item below):

☐ is attached hereto.☐ was filed as United States application

Serial No. _____

on _____

and was amended

on _____ (if applicable).

☒ was filed as PCT international applicationNumber PCT/CA00/00865on July 21, 2000

and was amended under PCT Article 19

on February 6, 2001 (if applicable).

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56(a).

I hereby claim foreign priority benefits under Title 35, United States Code, §119 of any foreign application(s) for patent or inventor's certificate or of any PCT international application(s) designating at least one country other than the United States of America listed below and have also identified below any foreign application(s) for patent or inventor's certificate or any PCT international application(s) designating at least one country other than the United States of America filed by me on the same subject matter having a filing date before that of the application(s) of which priority is claimed:

PRIOR FOREIGN/PCT APPLICATION(S) AND ANY PRIORITY CLAIMS UNDER 35 U.S.C. 119:

COUNTRY (If PCT indicate PCT)	APPLICATION NUMBER	DATE OF FILING (day, month, year)	PRIORITY CLAIMED UNDER 35 U.S.C. 119
Canada	2,280,434	August 18, 1999	<input checked="" type="checkbox"/> YES <input type="checkbox"/> NO
			<input type="checkbox"/> YES <input type="checkbox"/> NO
			<input type="checkbox"/> YES <input type="checkbox"/> NO
			<input type="checkbox"/> YES <input type="checkbox"/> NO
			<input type="checkbox"/> YES <input type="checkbox"/> NO

Combined Declaration for Patent Application and Power of Attorney (Continued)
(Includes Reference to PCT International Applications)ATTY'S DOCKET NUMBER
6670/OK215USO

I hereby claim the benefit under Title 35, United States Code, §120 of any United States application(s) or PCT international application(s) designating the United States of America that is/are listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in that/those prior application(s) in the manner provided by the first paragraph of Title 35, United States Code §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior application(s) and the national or PCT international filing date of this application:

PRIOR U.S. APPLICATIONS OR PCT INTERNATIONAL APPLICATIONS DESIGNATING THE U.S. FOR BENEFIT UNDER 35 U.S.C. 120:

U.S. APPLICATIONS		STATUS (Check one)		
U.S. APPLICATION NUMBER	U.S. FILING DATE	PATENTED	PENDING	ABANDONED
PCT APPLICATIONS DESIGNATING THE U.S.				
PCT APPLICATION NO	PCT FILING DATE	U.S. SERIAL NUMBER ASSIGNED (if any)		
PCT/CA00/00865	7/21/00			

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agents to prosecute this application and transact all business in the Patent and Trademark Office connected therewith. Michael J. Sweedler #19,937, S. Peter Ludwig #25,351, Paul Fields #20,298, Joseph B. Lerch #26,936, Melvin C. Garner #26,272, Ethan Horwitz #27,646, Adda C. Gogoris #29,714, Bert J. Lewen #19,407, Henry Sternberg #22,408, Peter C. Schechter #31,662, Robert Schaffer #31,194, Robert C. Sullivan, Jr. #30,499, Joseph R. Robinson #33,448, Paul F. Fehlner #35,135, Scott G. Lindvall #40,325, Pierre R. Yanney #35,418, and John C. Todaro #36,036.

Send Correspondence to:

John C. Todaro
DARBY & DARBY P.C.
805 Third Avenue
New York, New York 10022-7513

Direct Telephone Calls to:
(name and telephone number)

(212) 527-7700
John C. Todaro

2	FULL NAME OF INVENTOR	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
0	SCHULZ		Robert	
1	RESIDENCE & CITIZENSHIP	CITY	STATE OR FOREIGN COUNTRY	COUNTRY OF CITIZENSHIP
1	Sainte-Julie, Quebec		Canada CAX	Canada
2	POST OFFICE ADDRESS	POST OFFICE ADDRESS	CITY	STATE & ZIP CODE/COUNTRY
2	68 Bel Horizon		Sainte-Julie, Quebec	Canada JOL 2S0
2	FULL NAME OF INVENTOR	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
0	BOILY		Sabin	
2	RESIDENCE & CITIZENSHIP	CITY	STATE OR FOREIGN COUNTRY	COUNTRY OF CITIZENSHIP
2	Chambly, Quebec		Canada CAX	Canada
2	POST OFFICE ADDRESS	POST OFFICE ADDRESS	CITY	STATE & ZIP CODE/COUNTRY
2	1676 Labonte		Chambly, Quebec	Canada J3L 5M6
2	FULL NAME OF INVENTOR	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
0	DUBUC		Rene	
3	RESIDENCE & CITIZENSHIP	CITY	STATE OR FOREIGN COUNTRY	COUNTRY OF CITIZENSHIP
3	Sainte-Julie, Quebec		Canada CAX	Canada
2	POST OFFICE ADDRESS	POST OFFICE ADDRESS	CITY	STATE & ZIP CODE/COUNTRY
2	380 De Normandie		Sainte-Julie, Quebec	Canada J3E 1E8
2	FULL NAME OF INVENTOR	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
4	BLOUIN		Marco	
2	RESIDENCE & CITIZENSHIP	CITY	STATE OR FOREIGN COUNTRY	COUNTRY OF CITIZENSHIP
2	Montreal, Quebec		Canada CAX	Canada
2	POST OFFICE ADDRESS	POST OFFICE ADDRESS	CITY	STATE & ZIP CODE/COUNTRY
2	4601, 18e Avenue, app. 8		Montreal, Quebec	Canada H1X 2N6
2	FULL NAME OF INVENTOR	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
0	LALANDE		Guy	
5	RESIDENCE & CITIZENSHIP	CITY	STATE OR FOREIGN COUNTRY	COUNTRY OF CITIZENSHIP
5	Montreal, Quebec		Canada CAX	Canada
5	POST OFFICE ADDRESS	POST OFFICE ADDRESS	CITY	STATE & ZIP CODE/COUNTRY
5	358 St-Joseph Est, app. 1 8.L		Montreal, Quebec	Canada H2T 1J4

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patents issuing thereon.

SIGNATURE OF INVENTOR 201	SIGNATURE OF INVENTOR 202	SIGNATURE OF INVENTOR 203
<i>[Signature]</i>	<i>[Signature]</i>	<i>[Signature]</i>
DATE 9-04-2002	DATE 7-13-02	DATE 7-03-2002
SIGNATURE OF INVENTOR 204	SIGNATURE OF INVENTOR 205	
<i>[Signature]</i>	<i>[Signature]</i>	
DATE 7-03-2002	DATE 10-04-2002	